# **Material Safety Data Sheet**

Version 4.0 Revision Date 03/14/2010 Print Date 04/08/2010

# 1. PRODUCT AND COMPANY IDENTIFICATION

Product name : Iron(III) nitrate nonahydrate

Product Number : 529303 Brand : Aldrich

Company : Sigma-Aldrich Canada, Ltd

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CANADA

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# 2. HAZARDS IDENTIFICATION

# **Emergency Overview**

# **WHMIS Classification**

C Oxidizing Material Oxidizer

D2B Moderate skin irritant
Moderate respiratory irritant

Moderate respiratory irritant Moderate eye irritant

# GHS Label elements, including precautionary statements

Pictogram



Signal word Warning

Hazard statement(s)

H272 May intensify fire; oxidiser.
 H303 May be harmful if swallowed.
 H315 Causes skin irritation.
 H319 Causes serious eye irritation.
 H335 May cause respiratory irritation.

Precautionary statement(s)

P220 Keep/Store away from clothing/ combustible materials. P261 Avoid breathing dust/fume/gas/mist/vapours/spray.

P305 + P351 + P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if

present and easy to do. Continue rinsing.

**HMIS Classification** 

Health hazard: 2 Flammability: 0 Physical hazards: 1

### **Potential Health Effects**

Inhalation May be harmful if inhaled. Causes respiratory tract irritation.Skin May be harmful if absorbed through skin. Causes skin irritation.

**Eyes** Causes eye irritation.

**Ingestion** May be harmful if swallowed.

# 3. COMPOSITION/INFORMATION ON INGREDIENTS

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Synonyms : Ferric nitratenonahydrate

Formula :  $FeN_3O_9 \cdot 9H_2O$ Molecular Weight : 404.00 g/mol

CAS-No.	EC-No.	Index-No.	Concentration
Ferric nitrate nonahydrate			
7782-61-8	233-899-5	-	-

# 4. FIRST AID MEASURES

### General advice

Consult a physician. Show this safety data sheet to the doctor in attendance. Move out of dangerous area.

#### If inhaled

If breathed in, move person into fresh air. If not breathing give artificial respiration Consult a physician.

#### In case of skin contact

Wash off with soap and plenty of water. Consult a physician.

# In case of eye contact

Rinse thoroughly with plenty of water for at least 15 minutes and consult a physician.

### If swallowed

Never give anything by mouth to an unconscious person. Rinse mouth with water. Consult a physician.

### 5. FIRE-FIGHTING MEASURES

# Suitable extinguishing media

Use water spray, alcohol-resistant foam, dry chemical or carbon dioxide.

# Special protective equipment for fire-fighters

Wear self contained breathing apparatus for fire fighting if necessary.

### **Further information**

Use water spray to cool unopened containers.

### 6. ACCIDENTAL RELEASE MEASURES

# **Personal precautions**

Use personal protective equipment. Avoid dust formation. Avoid breathing dust. Ensure adequate ventilation. Evacuate personnel to safe areas.

### **Environmental precautions**

Do not let product enter drains.

# Methods and materials for containment and cleaning up

Pick up and arrange disposal without creating dust. Keep in suitable, closed containers for disposal.

# 7. HANDLING AND STORAGE

### Precautions for safe handling

Avoid contact with skin and eyes. Avoid formation of dust and aerosols.

Provide appropriate exhaust ventilation at places where dust is formed. Keep away from sources of ignition - No smoking. Keep away from combustible material.

### Conditions for safe storage

Keep container tightly closed in a dry and well-ventilated place.

hygroscopic Store under inert gas.

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### 8. EXPOSURE CONTROLS/PERSONAL PROTECTION

# Personal protective equipment

# Respiratory protection

Where risk assessment shows air-purifying respirators are appropriate use a full-face particle respirator type N100 (US) or type P3 (EN 143) respirator cartridges as a backup to engineering controls. If the respirator is the sole means of protection, use a full-face supplied air respirator. Use respirators and components tested and approved under appropriate government standards such as NIOSH (US) or CEN (EU).

### Hand protection

Handle with gloves.

# Eye protection

Safety glasses with side-shields conforming to EN166

# Skin and body protection

Choose body protection according to the amount and concentration of the dangerous substance at the work place.

# Hygiene measures

Handle in accordance with good industrial hygiene and safety practice. Wash hands before breaks and at the end of workday.

# 9. PHYSICAL AND CHEMICAL PROPERTIES

### **Appearance**

Form solid

# Safety data

pH no data available

Melting point 47 °C (117 °F) - lit.

Boiling point no data available

Flash point not applicable

Ignition temperature no data available

Lower explosion limit no data available

Upper explosion limit no data available

Density 1.68 g/cm3 at 20 °C (68 °F)

Water solubility soluble

# 10. STABILITY AND REACTIVITY

# Chemical stability

Stable under recommended storage conditions.

### Conditions to avoid

no data available

# Materials to avoid

Organic materials, Powdered metals

### Hazardous decomposition products

Hazardous decomposition products formed under fire conditions. - nitrogen oxides (NOx), Iron oxides

# 11. TOXICOLOGICAL INFORMATION

#### Acute toxicity

LD50 Oral - rat - 3,250 mg/kg

#### Skin corrosion/irritation

no data available

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# Serious eye damage/eye irritation

no data available

# Respiratory or skin sensitization

no data available

# Germ cell mutagenicity

no data available

### Carcinogenicity

IARC:

No component of this product present at levels greater than or equal to 0.1% is identified as probable, possible or confirmed human carcinogen by IARC.

# Reproductive toxicity

no data available

# Specific target organ toxicity - single exposure (GHS)

Inhalation - May cause respiratory irritation.

# Specific target organ toxicity - repeated exposure (GHS)

no data available

# **Aspiration hazard**

no data available

#### Potential health effects

**Inhalation** May be harmful if inhaled. Causes respiratory tract irritation.

**Ingestion** May be harmful if swallowed.

**Skin** May be harmful if absorbed through skin. Causes skin irritation.

**Eyes** Causes eye irritation.

# Signs and Symptoms of Exposure

Absorption into the body leads to the formation of methemoglobin which in sufficient concentration causes cyanosis. Onset may be delayed 2 to 4 hours or longer., Nausea, Dizziness, Headache, Weakness, Incoordination., Confusion., Cyanosis, Coma

# Additional Information

RTECS: NO7175000

# 12. ECOLOGICAL INFORMATION

# **Toxicity**

### Persistence and degradability

no data available

# Bioaccumulative potential

no data available

### Mobility in soil

no data available

# PBT and vPvB assessment

no data available

# Other adverse effects

no data available

### 13. DISPOSAL CONSIDERATIONS

### **Product**

Observe all federal, state, and local environmental regulations. Contact a licensed professional waste disposal service to dispose of this material.

# Contaminated packaging

Dispose of as unused product.

# 14. TRANSPORT INFORMATION

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DOT (US)

UN-Number: 1466 Class: 5.1

Proper shipping name: Ferric nitrate Reportable Quantity (RQ): 1000 lbs

Marine pollutant: No

Poison Inhalation Hazard: No

**IMDG** 

UN-Number: 1466 Class: 5.1

Proper shipping name: FERRIC NITRATE

Marine pollutant: No

**IATA** 

UN-Number: 1466 Class: 5.1

Proper shipping name: Ferric nitrate

Packing group: III

Packing group: III

Packing group: III

#### 15. REGULATORY INFORMATION

### **DSL Status**

All components of this product are on the Canadian DSL list.

# WHMIS Classification

C Oxidizing Material

D2B

Oxidizer
Moderate skin irritant
Moderate respiratory irritant
Moderate eye irritant

EMS-No: F-A, S-Q

# **16. OTHER INFORMATION**

#### **Further information**

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